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# Heath Laboratory Chemistry Experiments Answers

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This lab is an excerpt  
from Heath Chemistry,  
a textbook of  
laboratory  
experiments.

Information regarding  
the lab and detailed  
procedure are provided

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on the following pages.  
Note: Change Step 2  
from g. 0 g of copper  
(II) chloride to  
approximately 3.0 g -  
4.0 g

### **Lab 4B (student copy)**

To explain the  
observations obtained  
by applying Le  
Chatelier's Principle  
Apparatus and  
Materials: Refer to  
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Experiments page 209

Procedure: Refer to

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209-211 Data and

Observations: Part I

Table1: Equilibrium  
stresses involving

Thymol Blue Reagent

Added Stress (Ion

Added ...

**Lab 19A - Morgan  
Smith and Shaylin  
Quartly April  
8th, 2016 ...**

*Page 6/21*

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If the same thing  
happened in either  
experiment, CO<sub>2</sub> was  
produced by the

experiment. In the first  
experiment, Mg  
dissolves in HCl to  
make aqueous MgCl<sub>2</sub> --  
there should be no CO<sub>2</sub>  
liberated. In the...

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Experiment 1B -

Making Observations -



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Lab Safety Bingo: 3

.Chemistry 11 Lab:

Types of Chemical

Reactions (April . as

one of the four main

types H AZARDS Most

of these solutions are .

experiment

demonstrated the

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know how to make the  
table for question # 5.  
"From the results of  
the temperature at  
which a precipitate  
dissolved, make up a  
table showing the  
temperature, the  $K_{sp}$   
and the solubility."

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Lab Experiment 4b  
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Lab: 19 Purpose: To use and mix different

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concentration of solutions of  $\text{Pb}^{2+}$  and  $\text{I}^-$ , then to see if ppt is formed. To acquire the  $K_{sp}$  value of  $\text{PbI}_2$  at

room temperature and temperatures above the room temperature.

Materials: Please refer to p. 220 of Heath Chemistry Laboratory Experiments.

Procedure: 1. Obtain 12 test tubes and two test tube racks, then put 6 test tubes in each rack.



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**19c.docx - Lab 19  
Purpose To use and  
mix different ...**

Heath Chemistry Lab  
19A. Chemistry 12 -  
Lab 19A Flow Chart  
(Investigating Chemical  
Equilibrium) PART 1:  
Put on safety goggles  
and lab apron Obtain 2  
(Empty, Clean, AND  
DRY) 250mL  
Erlenmeyer flasks. Add  
approx. 100mL of  
DISTILLED water and  
1mL of thymol blue

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solution to each flask  
Record the colour of  
this solution in note  
book FIRST FLASK: add  
a single drop of 0.1M  
HCl.

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