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Answers States Of Matter  
Section 3

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## Answers States Of Matter **Modern Chemistry Chapter 10 Flashcards | Quizlet**

CHAPTER 10 REVIEW States of Matter  
SECTION 3 SHORT ANSWER Answer the following questions in the space provided. 1. Match description on the right to the correct crystal type on the left. b ionic crystal (a) has mobile electrons in the crystal c covalent molecular crystal (b) is hard, brittle, and nonconducting.

## **Modern Chemistry Chapter 10 Review Answer Key**

How It Works: Identify the lessons in Holt McDougal Modern Chemistry's States of Matter chapter with which you need help. Find the corresponding video lessons within this companion course chapter.

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## **Modern Chemistry Chapter 10 Review Worksheet Answers**

Modern Chemistry. Chapter 10. States of Matter. Kinetic-Molecular Theory The kinetic-molecular theory of matter is based on the idea that particles of matter are always in motion and this motion has consequences that affect its physical properties. Kinetic-Molecular Theory of Gases 1- Gases consist of large numbers of tiny particles that are far apart relative to their size. 2- Collisions between gas particles and between particles and their container walls are elastic.

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Review of Chemistry of Matter (Chapter 2, & 3 Test Review) ANSWERS Directions: Review and check all your answer with this answer sheet to get ready for the test on Chemistry of Matter (Chapter 2 & 3). Holt McDougal Modern Chemistry Chapter 10: States of Matter ...

## **Chemistry Chapter 10 Test Review Answers**

CHAPTER 10 REVIEW States of Matter  
SECTION 4 SHORT ANSWER Answer the following questions in the space provided. 1. a When a substance in a closed system undergoes a phase change and the system reaches equilibrium, (a) the two opposing changes occur at equal rates. (b) there are no more phase changes. (c) one

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phase change predominates.

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## **Modern Chemistry Chapter 10 Review Answers**

Name Date Class CHAPTER 10 REVIEW  
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ANSWER Answer the following questions  
in the space provided. 1. Identify  
whether the descriptions below describe  
an ideal gas or a real gas. ideal gas

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closed system undergoes a phase  
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